

Stabile, M. R., Hudlicky, T., Meisels, M. L., Butora, G., Gum, A. G., Fearnley, S. P., Thorpe, A. J. & Ellis, M. R. (1995). *Chirality*, **7**, 556–559.

*Acta Cryst.* (1998). **C54**, 1164–1165

## *N*-Benzyl-*N*-(*tert*-butyloxycarbonyl)-glycine, an *N*-Substituted Glycine (Peptoid) Monomer

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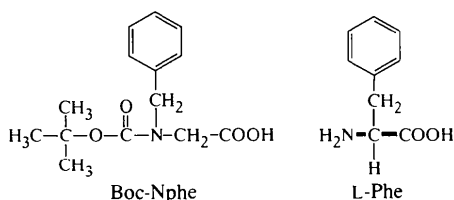
(Received 23 September 1997; accepted 12 February 1998)

### Abstract

The title compound, C<sub>14</sub>H<sub>19</sub>NO<sub>4</sub>, an amino acid mimic, was crystallized from ethyl acetate solution in a centrosymmetric space group. The distance between the side chain and the backbone was shorter than usually found in amino acids. The positional shift from  $\alpha$ -carbon to nitrogen produced no significant steric hindrance between the side chain and the *tert*-butyl group.

### Comment

Peptoids, amino acid mimics, have a basic *N*-substituted glycine unit and were designed as a new series of potentially bioactive compounds (Simon *et al.*, 1992; Zuckermann *et al.*, 1992). The side chains or functional groups are bonded to the  $\alpha$ -N atom, unlike the usual side chain, which is bound to the  $\alpha$ -C atom. The chemical structure is similar to that of a  $\beta$ -amino acid and, therefore, metabolic stability, reduction of conformational constraint by chirality and a wide variability of functional groups are expected (Figliozzi *et al.*, 1996). Relative to polypeptides, polypeptoids have their side chains shifted by one position along the backbone. A monomer derivative, *N*-benzyl-*N*-(*tert*-butyloxycarbonyl)glycine (Boc-Nphe), was crystallized from ethyl acetate solution in the centrosymmetric space group *P2*<sub>1</sub>/*a*.



The N1—C1B bond length is 1.456 (4) Å. As expected, this linkage is shorter than a C $\alpha$ —C $\beta$  bond length (1.54 Å). In comparison with the corresponding phenylalanine derivative, the benzyl group is spatially closer to the *tert*-butyl group by one covalent bond. No steric hindrance, however, was found in the title compound. The O5BT—C6BT—N1—C1A torsion angle, which corresponds to the  $\omega$  angle of a peptide bond, has a value of 177.4 (3)° and is in the *trans* region. Although the O5BT—C6BT—N1—C1B torsion angle in the peptoid is  $-1.3$  (3)°, no significant contact was found between the benzyl and *tert*-butyl groups. In packing, the molecules of Boc-Nphe form hydrogen-bonded dimers of O1T...O1 distance 2.622 (3) Å [O1T—H1T 0.819 (3), H1T...O1 1.807 (3) Å and O1T—H1T...O1 172.8 (2)°] across a center of symmetry at  $-x, -y, -z + 1$ .

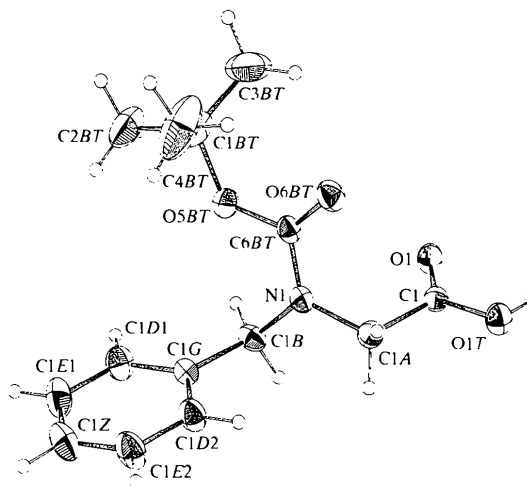


Fig. 1. A view of the title compound with displacement ellipsoids drawn at the 50% probability level.

### Experimental

The synthesis of the title compound was carried out according to Simon *et al.* (1992). A Schiff base was formed by mixing glyoxylic acid and benzylamine (molar ratio 1:1) in MeOH, and was hydrolyzed on Pd-carbon. The product was extracted with ethyl acetate (AcOEt) and aqueous NaHCO<sub>3</sub>, and was then reacted with di-*tert*-butyl dicarbonate in dioxane/aqueous NaOH solution. The reaction mixture was extracted with AcOEt and aqueous KHSO<sub>4</sub>. The AcOEt extract was condensed and the residue crystallized over a period of 2–3 d without any solvent.

### Crystal data

C<sub>14</sub>H<sub>19</sub>NO<sub>4</sub>  
M<sub>r</sub> = 265.30

Cu K $\alpha$  radiation  
 $\lambda$  = 1.54180 Å

## Monoclinic

 $P2_1/a$  $a = 13.323 (2) \text{ \AA}$  $b = 7.8005 (9) \text{ \AA}$  $c = 15.324 (2) \text{ \AA}$  $\beta = 110.499 (8)^\circ$  $V = 1491.7 (3) \text{ \AA}^3$  $Z = 4$  $D_x = 1.181 \text{ Mg m}^{-3}$  $D_m$  not measured

## Data collection

Rigaku AFC-5R diffractometer

 $2\theta - \omega$  scans

Absorption correction: none

2533 measured reflections

2413 independent reflections

1783 reflections with

 $I > 2\sigma(I)$ 

Cell parameters from 20 reflections

 $\theta = 19.80\text{--}20.03^\circ$  $\mu = 0.713 \text{ mm}^{-1}$  $T = 293 (2) \text{ K}$ 

Block

 $0.6 \times 0.4 \times 0.1 \text{ mm}$ 

Colorless

 $R_{\text{int}} = 0.018$  $\theta_{\text{max}} = 63.18^\circ$  $h = 0 \rightarrow 15$  $k = -9 \rightarrow 0$  $l = -17 \rightarrow 16$ 

3 standard reflections

every 100 reflections

intensity decay:  $-0.7\%$ 

## Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.068$  $wR(F^2) = 0.183$  $S = 1.176$ 

2365 reflections

176 parameters

H atoms constrained

 $w = 1/[\sigma^2(F_o^2) + (0.1062P)^2 + 0.6150P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\text{max}} < 0.001$  $\Delta\rho_{\text{max}} = 0.281 \text{ e \AA}^{-3}$  $\Delta\rho_{\text{min}} = -0.458 \text{ e \AA}^{-3}$ 

Extinction correction:

SHELXL93

Extinction coefficient:

0.0038 (8)

Scattering factors from

International Tables for Crystallography (Vol. C)

Scan widths were  $(1.628 + 0.3\tan\theta)^\circ$  in  $\omega$ , with a background/scan time ratio of 0.5. The data were corrected for Lorentz and polarization effects. The Laue group assignment, systematic absences and intensity statistics were consistent with centrosymmetric space group  $P2_1/a$ . Intensities were measured to the mechanical limit of the diffractometer; the  $\theta_{\text{max}}$  was set approximately at  $65^\circ$ . H atoms were calculated at idealized positions and refined with fixed isotropic displacement parameters ( $U_{\text{iso}} = 1.2U_{\text{eq}}$  for the associated C atom or  $1.5U_{\text{eq}}$  for methyl C atoms).

Data collection: *MSCIAFC Diffractometer Control Software* (Molecular Structure Corporation, 1991). Cell refinement: *MSCIAFC Diffractometer Control Software*. Data reduction: *MSCIAFC Diffractometer Control Software*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996). Software used to prepare material for publication: *PARST* (Nardelli, 1983).

This study was partially supported by the Grand-in-Aid for Scientific Research (07672427) from the Ministry of Education, Science, Sports and Culture, Japan.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SX1067). Services for accessing these data are described at the back of the journal.

## References

- Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Figliozzi, G. M., Goldsmith, R., Ng, S. C., Banville, S. C. & Zuckermann, R. N. (1996). *Methods Enzymol.* **267**, 437–447.
- Molecular Structure Corporation (1991). *MSCIAFC Diffractometer Control Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.
- Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. University of Göttingen, Germany.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Simon, R. J., Kania, R. S., Zuckermann, R. N., Huebner, V. D., Jewell, D. A., Banville, S., Ng, S., Wang, L., Rosenberg, S., Marlowe, C. K., Spellmeyer, D. C., Tan, R., Frankel, A. D., Santi, D. V., Cohen, F. E. & Bartlett, P. A. (1992). *Proc. Natl Acad. Sci. USA*, **89**, 9367–9371.
- Zuckermann, R. N., Kerr, J. M., Kent, S. B. H. & Moos, W. H. (1992). *J. Am. Chem. Soc.* **114**, 10646–10647.

*Acta Cryst.* (1998). **C54**, 1165–1168

## 9-Deoxy-15-hydroxy- and 9-Deoxy-19-hydroxycotlenol†

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(Received 2 December 1997; accepted 11 February 1998)

## Abstract

The title analogs (both  $C_{21}H_{34}O_4$ ) of cotlenol, a plant-growth regulator, both have a chair–sofa eight-membered ring, which has been recognized as important for the biological activity of this class of compounds.

## Comment

Cotlenol, (I) (Sassa *et al.*, 1975), is a common aglycon of cotlenins and is known to have potent plant hormone-like activity, similar to fusicoccin. Since the binding protein of fusicoccin has recently been identified as a member of the 14–3–3 proteins (Korthout & De Boer, 1994), these fusicoccane diterpenoids have at-

† Alternative nomenclature: (1R,3aS,4R,5S,9aR)-1,2,3,3a,4,5,6,8,9,9a-decahydro-7-(1-hydroxy-1-methylethyl)-1-(methoxymethyl)-4,9a-dimethyldicyclopenta[*a,d*]cyclooctene-1,5-diol and (1R,3aS,4R,5S,9aR)-1,2,3,3a,4,5,6,8,9,9a-decahydro-7-[(S)-2-hydroxy-1-methylethyl]-1-(methoxymethyl)-4,9a-dimethyldicyclopenta[*a,d*]cyclooctene-1,5-diol.